



Projekt współfinansowany przez



	KAPITAŁ LUDZKI NARODOWA STRATEGIA SPÓJNOŚCI	Unię Europe Europejskie	jską w ramach EUROPEJSKI ego Funduszu FUNDUSZ SPOŁECZNY			
Course title			ECTS code			
Catalysis in inorganic chemistry			13.3.1204			
Name of unit admir			10:0.1204			
null						
Studies						
foculty	field of study	tupo	type pierwszego stopnia			
faculty Wydział Chemii	field of study Biznes chemiczny		form stacjonarne			
			specialty wszystkie			
			cialization wszystkie			
Wydział Chemii	Chemia		type pierwszego stopnia			
			form stacjonarne specialty chemia biomedyczna, chemia kosmetyków, analityka i diagnostyka			
		specially	chemiczna, chemia żywności			
		specialization	cialization wszystkie			
Wydział Chemii	Ochrona środowiska		type pierwszego stopnia			
			form stacjonarne			
			specialty wszystkie			
		specialization	WSZYSTKIE			
dr hab. Dagmara Jacewicz, profesor uczelni; dr Joann Forms of classes, the realization and number of hour			ECTS credits			
Forms of classes			2			
Laboratory classes, Lecture			classes - 15 h			
The realization of activities			tutorial classes - 15 h			
			studnet's own work - 20 h			
classroom instruction  Number of hours			TOTAL: 50 h - 2 ECTS			
			1011210011 2 2010			
	Laboratory classes: 11 hours					
The academic cycl	е					
2025/2026 summ	er semester					
Type of course		Langua	Language of instruction			
an elective course	9	Englis	English			
Teaching methods		examina	Form and method of assessment and basic criteria for eveluation or examination requirements			
- Lecture based on the multimedia presentation			Final evaluation			
	ory work - chemical experiment	ıs,   Grade	Graded credit			
analysis of obtai	ned results and discussion.		Assessment methods			
			re, Laboratory classes – short test and report from performed			
			chemical experiments.  The basic criteria for evaluation			
		The bas	ic criteria for evaluation			

Sylabusy - Centrum Informatyczne UG Dział Kształcenia



Lecture and Laboratory classes: positive note from an short test with 5 open questions:

91-100% 5.0 81-90% 4.5 71-80% 4.0 61-70% 3.5 51-60% 3.0 < 51% 2.0

- a. passing short test covering the material of the Lecture and Laboratory classes
- assessment criteria in accordance with the University of Gdańsk
   Study Regulations

## Method of verifying required learning outcomes

#### Required courses and introductory requirements

## A. Formal requirements

lack

### B. Prerequisites

lack

### Aims of education

- familiarize students with the main aspects of mechanisms of action of catalysts in inorganic chemistry
- familiarize students with the application of catalysts in inorganic chemistry in practice
- presenting the homogeneous and heterogeneous catalysts
- familiarize students with the methods of synthesis of homogeneous and heterogeneous catalysts in inorganic chemistry

#### **Course contents**

The aim of the course is to understand of the mechanisms of action of catalysts in inorganic chemistry, as well as to learn their application in practice. Laboratory exercises include learning and synthesis of homogeneous and heterogeneous catalysts in inorganic chemistry, conducting catalytic reactions with the participation of selected catalysts and inhibitors, and also evaluation of their operation under the influence of changing reaction conditions.

## Bibliography of literature

Literature required to pass the course

- D. Astruc, "Organometallic Chemistry and Catalysis", Springer, 2007.
- R. van Eldik, C. Hubbard, "Advances in Inorganic Chemistry" vol. 65, Elsevier, 2013.
- C. Housecroft, "Inorganic Chemistry", Pearson, 2018. Extracurricular readings
- M. A. Vannice "Kinetics of Catalytic Reaction", Springer, 2005.

# The learning outcomes (for the field of study and specialization)

### Chemical Business:

K\_BCh\_W02

enumerates laws and theories in chemistry, physics and mathematics necessary to formulate and solve simple engineering tasks

K\_BCh\_U01

on the basis of the acquired knowledge, identifies, analyses and solves engineering tasks and problems in broadly understood chemistry

K\_BCh\_U08

uses the chemical nomenclature and engineering terminology properly

K\_BCh\_K02

works individually demonstrating initiative and independence in actions, and effectively cooperates in a team, performing various roles in it

## Chemistry:

K W02

describes in an advanced level the properties of elements

#### Knowledge

Student:

- knows chemical laws relating to chemical compounds (catalytic control chemical transformations)
- knows the chemical nomenclature used to describe chemical compounds (homogeneous and heterogeneous catalysis)
- understands the role of activation energy in catalytic chemical reactions
- knows the methods of testing the effectiveness of catalysts (turnover frequency and turnover number)
- knows the methods of catalysts preparation and methods of their physicochemical analysis
- understands the need to conduct an experiment in order to obtain empirical data to explain the processes taking place

#### **Skills**

#### Student:

- is active in planning an experiment regarding catalytic reactions
- is able to synthesize catalysts and inhibitors for the needs of inorganic chemistry reactions
- is able to draw conclusions from the conducted experiments, e.g. calculate turnover frequency and turnover number of catalysts
- is able to synthesize catalysts on inorganic carriers



and the most important chemical compounds, enumerates the methods of their preparation and methods of analysis

performs analyses using experimental methods and draws conclusions based on them

K U04

plans and performs chemical experiments and analyses the results obtained

K U09

is able to learn independently

K\_K05

observes established procedures in laboratory work and is responsible for the safety of her/his and others' work

**Environmental Protection:** 

K OŚI W04

explains at an advanced level the meaning and indispensability of empirical data in the description and interpretation of natural phenomena and processes (occurring in the environment)

K\_OŚI\_U02

plans, selects appropriate research and measuring equipment and devices, performs physicochemical measurements and experiments; analyses the results and draws conclusions based on them

K OŚI K02

works individually demonstrating initiative and independence in actions, and effectively cooperates in a team, performing various roles in it

K\_OŚI\_K05

identifies the level of her/his knowledge and skills, demonstrates the need to update knowledge about the environment and its protection, demonstrates the need for continuous professional training and personal development - is able to carry out reactions with the participation of a catalyst and an inhibitor as well as to optimize process conditions

## Social competence

#### Student:

- engages in scientific discussions in group
- is active in extending knowledge and understands the need for continuous increasing the level of knowledge and qualifications
- is able to work in a safe way for all participants of the class group
- $\mbox{-}$  can play various roles in the group when solving research problems and performing experiments

# Catalysis in inorganic chemistry #13.3.1204

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