


**KAPITAŁ LUDZKI**  
 NARODOWA STRATEGIA SPÓJNOŚCI

 Projekt współfinansowany przez  
 Unię Europejską w ramach  
 Europejskiego Funduszu  
 Społecznego

**UNIA EUROPEJSKA**  
 EUROPEJSKI  
 FUNDUSZ SPOŁECZNY


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|---|--------------------|---|-------------------------|
| <b>Nazwa przedmiotu</b>   |                    | <b>Kod ECTS</b>   |                         |
| Principles and applications of fluorescence spectroscopy  |                    | 13.3.1222   |                         |
| <b>Nazwa jednostki prowadzącej przedmiot</b>  |                    |   |                         |
| Katedra Chemii Ogólnej i Nieorganicznej   |                    |   |                         |
| <b>Studia</b>   |                    |   |                         |
| <b>wydział</b>  | <b>kierunek</b>    | <b>poziom</b>   | <b>drugiego stopnia</b> |
| Wydział Chemii  | Biznes chemiczny   | <b>forma</b>  | stacjonarne             |
|   |                    | <b>moduł specjalnościowy</b>  | wszystkie               |
|   |                    | <b>specjalizacja</b>  | wszystkie               |
| Wydział Chemii  | Chemia             | <b>poziom</b>   | drugiego stopnia        |
|   |                    | <b>forma</b>  | stacjonarne             |
|   |                    | <b>moduł specjalnościowy</b>  | wszystkie               |
|   |                    | <b>specjalizacja</b>  | wszystkie               |
| Wydział Chemii  | Ochrona środowiska | <b>poziom</b>   | drugiego stopnia        |
|   |                    | <b>forma</b>  | stacjonarne             |
|   |                    | <b>moduł specjalnościowy</b>  | wszystkie               |
|   |                    | <b>specjalizacja</b>  | wszystkie               |
| <b>Nazwisko osoby prowadzącej (osób prowadzących)</b>   |                    |   |                         |
| dr inż. Krzysztof Żamojć  |                    |   |                         |
| <b>Formy zajęć, sposób ich realizacji i przypisana im liczba godzin</b>   |                    | <b>Liczba punktów ECTS</b>  |                         |
| <b>Formy zajęć</b>  |                    | 4   |                         |
| Wykład, Ćw. laboratoryjne   |                    | classes - 30 h  |                         |
| <b>Sposób realizacji zajęć</b>  |                    | tutorial classes - 30 h   |                         |
| zajęcia w sali dydaktycznej   |                    | student's own work - 40 h   |                         |
| <b>Liczba godzin</b>  |                    | TOTAL: 100 h - 4 ECTS   |                         |
| Ćw. laboratoryjne: 15 godz., Wykład: 15 godz.   |                    |   |                         |
| <b>Termin realizacji przedmiotu</b>   |                    |   |                         |
| 2023/2024 zimowy  |                    |   |                         |
| <b>Status przedmiotu</b>  |                    | <b>Język wykładowy</b>  |                         |
| fakultatywny (do wyboru)  |                    | angielski   |                         |
| <b>Metody dydaktyczne</b>   |                    | <b>Forma i sposób zaliczenia oraz podstawowe kryteria oceny lub wymagania egzaminacyjne</b> |                         |
| <ul style="list-style-type: none"> <li>- Lecture with the use of the multimedia presentation on fluorescence spectroscopy</li> <li>- Practical laboratory work – chemical experiments, analysis of obtained results and discussion</li> </ul> |                    | <b>Sposób zaliczenia</b>  |                         |
|   |                    | Zaliczenie na ocenę   |                         |
|   |                    | <b>Formy zaliczenia</b>   |                         |
|   |                    | Lecture – exam with open and closed questions   |                         |
|   |                    | Laboratory classes – reports and short tests  |                         |
|   |                    | <b>Podstawowe kryteria oceny</b>  |                         |

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|  | <p>Lecture: a positive note from an exam with approximately 20 open and closed questions:</p> <p>91-100%: 5.0<br/>81-90%: 4.5<br/>71-80%: 4.0<br/>61-70%: 3.5<br/>51-60%: 3.0<br/>&lt; 51%: 2.0</p> <p>Laboratory classes: a positive note from all short tests and reports; final note is an average from notes from tests and reports:</p> <p>91-100%: 5.0<br/>81-90%: 4.5<br/>71-80%: 4.0<br/>61-70%: 3.5<br/>51-60%: 3.0<br/>&lt; 51%: 2.0</p>  |
| <b>Sposób weryfikacji założonych efektów uczenia się</b>   |   |
| <p>The method of verifying the acquisition of knowledge:<br/>Written exam in the field of fluorescence spectroscopy. During the lecture and laboratory exercises, the student solves problems in writing (short tests, reports) or oral (oral answer) in the field of fluorescence spectroscopy.</p> <p>The method of verifying the acquisition of skills:<br/>Assessment of the student's involvement in discussions on the issues related to the subject. Assessment of independent conducting of chemical experiments by the student. Assessment of the Student's explanation of the course of chemical experiments, assessment of the correctness of the analysis of the results, drawing conclusions from the experiments and preparation of reports.</p> <p>The method of verifying the acquisition of social competences:<br/>Assessment of the student's ability to solve scientific and research problems on the basis of individual and team work.</p> |   |
| <b>Określenie przedmiotów wprowadzających wraz z wymogami wstępnymi</b>  |   |
| <p><b>A. Wymagania formalne</b><br/>lack</p> <p><b>B. Wymagania wstępne</b><br/>lack</p>   |   |
| <b>Cele kształcenia</b>  |   |
| <p>Familiarize students with the basic and more advanced aspects of fluorescence spectroscopy.<br/>Familiarize students with the use of spectrofluorometer.</p>  |   |
| <b>Treści programowe</b>   |   |
| <p>Topics of the lecture: absorption of the light; ground and excited states; types of electronic transitions; absorption, excitation, and emission spectra; photophysical processes in the excited state; fluorescence quenching; the studies of the mechanisms of reactions - qualitative and quantitative methods; instrumentation.</p> <p>Topics of laboratory classes: spectrofluorometer operation; basic definitions and laws related with fluorescence spectroscopy; the studies of the mechanisms of fluorescence quenching; determination of aggregation number, cmc and hydrophobicity of the surfactants' micelles with the use of steady-state fluorescence emission spectra; the studies of the influence of various factors on the fluorescence emission spectra; determination of the stoichiometry and association constants of complexes.</p>  |   |
| <b>Wykaz literatury</b>  |   |
| <p>Extracurricular readings<br/>J.R. Lakowicz – Principles of fluorescence spectroscopy<br/>B. Valeur – Molecular fluorescence</p>   |   |
| <b>Kierunkowe efekty uczenia się</b><br><br>Chemical Business:<br>K_BChII_W06<br>knows and understands tasks in the field of chemistry, environmental protection and economics that are the subject of human activity to a degree that allows independent work on a research, scientific and measurement position<br>K_BChII_U04<br>is able to independently plan and perform specific research  | <b>Wiedza</b><br><br>Students: know and interpret main photophysical processes in ground and excited states as well as basic definitions and laws related with fluorescence spectroscopy; define types of electronic transitions; know differences and similarities between absorption, excitation and emission spectra; know the definitions and main mechanisms of fluorescence quenching; know main qualitative and quantitative methods of the studies of mechanisms or reactions; know the build of spectrofluorometer; define various factors which have an |

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| <p>tasks in the field or in the laboratory, interpret their results working individually or in a team, assuming various roles and functions in it</p> <p>K_BChII_K04</p> <p>is willing to properly assess the acquired knowledge, respect it and disseminate it in order to solve specific cognitive and practical issues</p> <p>Chemistry:</p> <p>K_W01</p> <p>uses in-depth knowledge of spectroscopic methods of chemical compound analysis</p> <p>K_U01</p> <p>plans and implements chemical experiments of extended complexity</p> <p>K_U02</p> <p>critically assesses the results of conducted, performed observations and theoretical calculations and discusses errors</p> <p>K_U04</p> <p>applies acquired knowledge of chemistry and related scientific disciplines</p> <p>K_K02</p> <p>works in a team taking on various roles in it</p> <p>Environmental Protection:</p> <p>K_OŚII_W09</p> <p>applies safety and hygiene principles when working independently on a test or measurement stand in a laboratory or in the field</p> <p>K_OŚII_U07</p> <p>has advanced skills in presenting the results of their own research, discussions based on literature data and public speaking, including leading a debate</p> <p>K_OŚII_K02</p> <p>recognises threats, creates safe work conditions and is responsible for the safety of own and other people's work</p> <p>K_OŚII_K07</p> <p>is willing to undertake individual and team activity; to professionally plan and organise its course and set priorities for their actions</p> | <p>influence on the fluorescence emission spectra.</p> <p><b>Umiejętności</b></p> <p>Students: present plainly – in both speech and writing – correct chemical argumentation, interpret and analyze information connected with fluorescence spectroscopy presented as text, tables, plots, schemes, figures, can use spectrofluorometer, can experimentally determine the mechanism of fluorescence quenching, aggregation number, cmc and hydrophobicity of the surfactants' micelles, can determine the influence of various factors on the fluorescence emission spectra as well as the stoichiometry and association constants of complexes, interpret information, formulate conclusions and explain opinions.</p> <p><b>Kompetencje społeczne (postawy)</b></p> <p>Students: understand need for learning, inspire other for learning; cooperate in group, taking different roles; exhibit creativity in determination of priorities necessary for realization of different tasks; understand social aspects of practical use of knowledge and abilities as well as connected with them responsibility.</p> |
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