



KAPITAŁ LUDZKI
NARODOWA STRATEGIA SPÓJNOŚCI

Projekt współfinansowany przez
Unię Europejską w ramach
Europejskiego Funduszu
Społecznego

UNIA EUROPEJSKA
EUROPEJSKI
FUNDUSZ SPOŁECZNY



Course title		ECTS code	
Principles and applications of fluorescence spectroscopy		13.3.1222	
Name of unit administrating study			
Faculty of Chemistry			
Studies			
faculty	field of study	type	second tier studies (MA)
Faculty of Chemistry	Chemical Business	form	full-time
		specialty	all
		specialization	all
Faculty of Chemistry	Chemistry	type	second tier studies (MA)
		form	full-time
		specialty	all
Faculty of Chemistry	Environmental Protection	specialization	all
		type	second tier studies (MA)
		form	full-time
		specialty	all
		specialization	all
Teaching staff			
dr Krzysztof Żamojć			
Forms of classes, the realization and number of hours		ECTS credits	
Forms of classes		4 classes - 30 h tutorial classes - 30 h student's own work - 40 h TOTAL: 100 h - 4 ECTS	
Laboratory classes, Lecture			
The realization of activities			
lectures in the classroom			
Number of hours			
Lecture: 15 hours, Laboratory classes: 15 hours			
The academic cycle			
2022/2023 winter semester			
Type of course		Language of instruction	
an elective course		english	
Teaching methods		Form and method of assessment and basic criteria for evaluation or examination requirements	
<ul style="list-style-type: none"> - Lecture with the use of the multimedia presentation on fluorescence spectroscopy; - Practical laboratory work – chemical experiments, analysis of obtained results and discussion 		Final evaluation	
		Graded credit	
		Assessment methods	
		Lecture – exam with open and closed questions Laboratory classes – reports and short tests	
		The basic criteria for evaluation	

Lecture: a positive note from an exam with approximately 20 open and closed questions:
91-100%: 5.0
81-90%: 4.5
71-80%: 4.0
61-70%: 3.5
51-60%: 3.0
< 51%: 2.0
Laboratory classes: a positive note from all short tests and reports; final note is an average from notes from tests and reports:
91-100%: 5.0
81-90%: 4.5
71-80%: 4.0
61-70%: 3.5
51-60%: 3.0
< 51%: 2.0

Method of verifying required learning outcomes

The method of verifying the acquisition of knowledge:

Written exam in the field of fluorescence spectroscopy. During the lecture and laboratory exercises, the student solves problems in writing (short tests, reports) or oral (oral answer) in the field of fluorescence spectroscopy.

The method of verifying the acquisition of skills:

Assessment of the student's involvement in discussions on the issues related to the subject. Assessment of independent conducting of chemical experiments by the student. Assessment of the Student's explanation of the course of chemical experiments, assessment of the correctness of the analysis of the results, drawing conclusions from the experiments and preparation of reports.

The method of verifying the acquisition of social competences:

Assessment of the student's ability to solve scientific and research problems on the basis of individual and team work.

Required courses and introductory requirements

A. Formal requirements

lack

B. Prerequisites

lack

Aims of education

Familiarize students with the basic and more advanced aspects of fluorescence spectroscopy.

Familiarize students with the use of spectrofluorometer.

Course contents

Topics of the lecture: absorption of the light; ground and excited states; types of electronic transitions; absorption, excitation, and emission spectra; photophysical processes in the excited state; fluorescence quenching; the studies of the mechanisms of reactions - qualitative and quantitative methods; instrumentation.

Topics of laboratory classes: spectrofluorometer operation; basic definitions and laws related with fluorescence spectroscopy; the studies of the mechanisms of fluorescence quenching; determination of aggregation number, cmc and hydrophobicity of the surfactants' micelles with the use of steady-state fluorescence emission spectra; the studies of the influence of various factors on the fluorescence emission spectra; determination of the stoichiometry and association constants of complexes.

Bibliography of literature

Extracurricular readings

J.R. Lakowicz – Principles of fluorescence spectroscopy

B. Valeur – Molecular fluorescence

The learning outcomes (for the field of study and specialization)

Chemical Business:

K_BChII_W06

knows and understands tasks in the field of chemistry, environmental protection and economics that are the subject of human activity to a degree that allows independent work on a research, scientific and measurement position

K_BChII_U04

Knowledge

Students: know and interpret main photophysical processes in ground and excited states as well as basic definitions and laws related with fluorescence spectroscopy; define types of electronic transitions; know differences and similarities between absorption, excitation and emission spectra; know the definitions and main mechanisms of fluorescence quenching; know main qualitative and quantitative methods of the studies of mechanisms or reactions; know the build of spectrofluorometer; define various factors which have an

<p>is able to independently plan and perform specific research tasks in the field or in the laboratory, interpret their results working individually or in a team, assuming various roles and functions in it K_BChII_K04</p> <p>is willing to properly assess the acquired knowledge, respect it and disseminate it in order to solve specific cognitive and practical issues Chemistry: K_W01</p> <p>uses in-depth knowledge of spectroscopic methods of chemical compound analysis K_U01</p> <p>plans and implements chemical experiments of extended complexity K_U02</p> <p>critically assesses the results of conducted, performed observations and theoretical calculations and discusses errors K_U04</p> <p>applies acquired knowledge of chemistry and related scientific disciplines K_K02</p> <p>works in a team taking on various roles in it Environmental Protection: K_OŚII_W09</p> <p>applies safety and hygiene principles when working independently on a test or measurement stand in a laboratory or in the field K_OŚII_U07</p> <p>has advanced skills in presenting the results of their own research, discussions based on literature data and public speaking, including leading a debate K_OŚII_K02</p> <p>recognises threats, creates safe work conditions and is responsible for the safety of own and other people's work K_OŚII_K07</p> <p>is willing to undertake individual and team activity; to professionally plan and organise its course and set priorities for their actions</p>	<p>influence on the fluorescence emission spectra.</p> <p>Skills</p> <p>Students: present plainly – in both speech and writing – correct chemical argumentation, interpret and analyze information connected with fluorescence spectroscopy presented as text, tables, plots, schemes, figures, can use spectrofluorometer, can experimentally determine the mechanism of fluorescence quenching, aggregation number, cmc and hydrophobicity of the surfactants' micelles, can determine the influence of various factors on the fluorescence emission spectra as well as the stoichiometry and association constants of complexes, interpret information, formulate conclusions and explain opinions.</p> <p>Social competence</p> <p>Students: understand need for learning, inspire other for learning; cooperate in group, taking different roles; exhibit creativity in determination of priorities necessary for realization of different tasks; understand social aspects of practical use of knowledge and abilities as well as connected with them responsibility.</p>
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